The coordination chemistry of iminooxosulphuranes

II*. The interaction of tosyl-NSO with osmium-alkylidene complexes

Max Herberhold * and Anthony F. Hill

Laboratorium für Anorganische Chemie der Universität Bayreuth, Universitätsstrasse 30, D-8580 Bayreuth (F.R.G.)

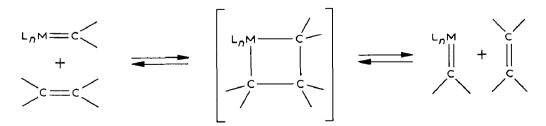
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Abstract

The low-valent osmium alkylidene complexes $Os(=CHR)Cl(NO)(PPh_3)_2$ (R = H, CO_2Et) react with 4-tosyliminooxosulphurane, 4-MeC_6H_4SO_2N=S=O, to give the four-membered metallacyclic complexes $Os[CHRS(=N-SO_2C_6H_4Me-4)O]Cl(NO)-(PPh_3)_2$ which provide models for intermediates proposed in the pseudo-Wittig synthesis of alkylideniminosulphuranes from iminooxosulphuranes and phosphorus ylides.

Introduction

The transition metal alkylidene catalysed metathesis of olefins is now a well understood and increasingly exploited reaction [2,3]. Fundamental to the currently accepted mechanism is the formation of a four-membered metallacycle derived from the cycloaddition of M=C and C=C double bonds [4].



* For Part I see ref. 1.

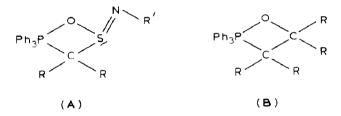
Surprisingly in view of the current interest in isolobal mappings [5], the extension of this reaction to the synthesis of more exotic ring systems, in particular inorganic analogues, has received little attention. Roper et al. [6] have described the interaction of a low-valent osmium methylene complex with sulphur dioxide to provide a complex of sulphene, $[CH_2=SO_2]$. We describe here the reaction of osmium al-kylidene complexes with iminooxosulphuranes, sulphur(IV) heterocumulenes in which one oxygen atom of SO₂ has been replaced by an imino group.

Results and discussion

The synthesis of alkylideniminosulphuranes, $R_2C=S=NR'$, by an inorganic "pseudo-Wittig" reaction has been reported [7]:

$$R' = N = 0$$
 + $PPh_3 = CR_2$ + $Ph_3P = 0$

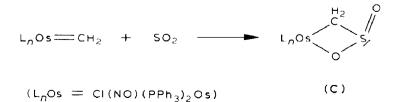
An attractive intermediate in this process is the heterocycle A, by analogy with the Wittig intermediate B.



A transition metal alkylidene complex may, depending upon the nature of metal and alkylidene substitution, display ylidic character, as, e.g., in Ta(=CH⁺Bu)(CH₂-⁺Bu)₃ [8], Ta(=CH₂)(CH₃)(η^5 -C₅H₅)₂ [9] and Os(=CH₂)Cl(NO)(PPh₃)₂ [10], and may act as an alkylidene transfer reagent in a manner reminiscent of the Wittig reaction [11].

L_nM=CR₂ + O=CR'₂ → 1/m(L_nM=O)_m + R₂C=CR'₂
e.g. L_nM=CR₂ = (^tBuCH₂)₃Ta=CH^tBu [cf. Ref. 12]
= "(
$$\eta^{5}$$
-C₅H₅)₂Ti=CH₂" [cf. Ref. 13]

These reactions presumably proceed through four-membered ring species like A and **B**, isolobally related to the olefin metathesis intermediate. An indication that main group multiple-bond systems could also participate in similar reactions was provided by the reaction between $Os(=CH_2)Cl(NO)(PPh_3)_2$ and sulphur dioxide to give the cyclic adduct C [6]:



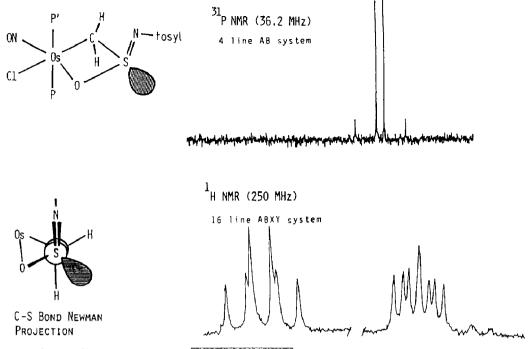
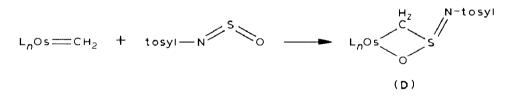


Fig. 1. ¹H and ³¹P NMR spectra of Os[CH₂S(=N-tosyl)-O]Cl(NO)(PPh₃)₂.

It was therefore reasonable to expect that replacing SO_2 with iminooxosulphuranes might lead to stable metallacyles isolobal with A. Indeed, whilst simple iminooxosulphuranes (MeC₆H₄NSO, ¹BuNSO) fail to provide the desired products (C being the only isolated complex), the highly electrophilic tosyliminooxosulphurane, MeC₆H₄SO₂NSO, reacts smoothly with Os(=CH₂)Cl(NO)(PPh₃)₂ to provide the cycloadduct **D**:



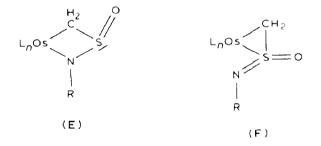
The asymmetry of the chiral sulphur atom in **D** imposes chemical inequivalence upon both the methylenic protons and the phosphane phosphorus atoms, giving rise to an ABXY spin system, manifest in the ${}^{31}P{}^{1}H$ NMR spectrum as an AB quartet and in the ${}^{1}H$ NMR spectrum as a 16-line pattern (Fig. 1).

The infrared spectrum of **D** features a strong nitrosyl stretching absorption at 1799 cm⁻¹, a value typical of divalent osmium, and two structured bands at ca. 1300 and 1144 cm⁻¹ which may be assigned to the SO₂ group of the toluene sulphonyl substituent. The remaining infrared bands not attributable to ligands present in the precursor methylene complex, appearing at 1026, 1008 and 661 cm⁻¹, are difficult to assign, and certainly do not unequivocally discriminate between alternative formulations of the adduct "Os(=CH₂)Cl(NO)(PPh₃)₂/(tosyl-NSO)" given by **E** and **F**. Accordingly, an X-ray diffraction study was performed on

$\frac{\text{Complex}}{\text{R} = \text{SO}_2\text{C}_6\text{H}_4\text{Me-4}}$	Infrared (Nujol) (cm ⁻¹)		NMR (ppm)	
	ν(NO)	Other	$\delta(^1H)^{a}$	$\delta({}^{31}\mathbf{P})({}^{1}\mathbf{H}){}^{b}$ [${}^{2}J(\mathbf{P},\mathbf{P}](\mathbf{H}z)$
$\overline{Os[CH_2S(=O)O]Cl(NO)(PPh_3)_2}$ (C) (ref. 6)	1797	1120, 1100, 709	3.01(m)	-2.211.0 [382]
((c), 0)	1/3/	1120, 1100, 707		· 2.2, = 11.0 [502]
$Os[CH_2S(=NR)O]Cl(NO)(PPh_3)_2$ (D)	1799	1309, 1299, 1284,	3.77 °	-0.4, -9.9 [369]
		1144, 1026, 1008,	3.16 ^d	
		661	2.37(s)	
$O_{s}[CH(CO_{2}Et)S(=NR)O]Cl(NO)(PPh_{3})_{2}$	1794	1283, 1143, 1023,	$6.02(t)^{-e}$	-4.4, 0.0 [371]
		990, 813, 791	3.67(q) /	-
			2.37(s)	
			$1.00(t)^{-f}$	

^{*a*-1}H NMR data determined from saturated solutions in CDCl₃ at 25 °C (250 MHz) relative to internal SiMe₄. ^{*b*} Determined at 25 °C in CDCl₃ solution relative to external D_3PO_4 . ^{*c*-3}*J*(P,H) 7.2, 9.0 Hz; ²*J*(H,H) 8.2 Hz. ^{*d*-3}*J*(P,H) 3.2, 5.6 Hz; ²*J*(H,H) 8.2 Hz. ^{*c*-3}*J*(P,H) 7.3 Hz. ^{*f*-3}*J*(H,H) 7.2 Hz.

crystals of the adduct [14]. Unfortunately, due to a disorder problem arising from the aplanarity of the metallacyclic ring involving the large tosyl substituent, the structure determination could not be taken to completion. The preliminary results indicate, however, that the metallacycle D is indeed formed, and that an exocyclic tosylimino group is present.



The carboethoxymethylene complex $Os(=CHCO_2Et)Cl(NO)(PPh_3)_2$ [15] prepared from $Os(CH_2CH_2)Cl(NO)(PPh_3)_2$ and ethyl diazoacetate, also reacts with tosyliminooxosulphurane to provide an adduct with spectroscopic data similar to **D** (Table 1). Only one isomer is formed (${}^{31}P{}^{1}H{}$ NMR) and, assuming steric factors to be dominant in the congested coordination sphere, an antiperiplanar disposition of tosylimino and carboethoxy groups is most likely.

Experimental

General experimental techniques and instrumentation have been described previously [1]. The compounds $Os(=CH_2)Cl(NO)(PPh_3)_2$ [10], $Os(CH_2CH_2)-Cl(NO)(PPh_3)_2$ [16] and tosyliminooxosulphurane [17] were prepared by published procedures. All manipulations were carried out under purified nitrogen. Characteris-

Table 1

Spectroscopic data for the osmium complexes

tic data for the new cycloadducts are compiled in Table 1 in addition to those for the related complex $Os[CH_2S(=O)O]Cl(NO)(PPh_3)_2$ (C) [6].

Preparation of the cycloadducts

 $Os[CH_2S(=NSO_2C_6H_4Me^{-4})O]Cl(NO)(PPh_3)_2$ (**D**). A suspension of Os(=CH₂)Cl(NO)(PPh₃)₂ (0.40 g, 0.50 mmol) in toluene (20 cm³) was treated with a solution of tosyliminooxosulphurane in toluene (0.10 mol/1, 5.3 ml, 0.53 mmol) and stirred for 2 h. The solvent was removed in vacuo and the yellow residue recrystallised from dichloromethane/ethanol as yellow needles. Yield 0.41 g (81%). M.p. 212°C. Anal. Found: C, 51.68; N, 2.85. C₄₄H₃₉ClN₂O₄OsP₂S₂ calc: C, 52.25; N, 2.77%. The complex was also characterised by a partially resolved crystallographic structural determination [14].

 $Os[CH_{2}CO_{2}C_{2}H_{5})S(=NSO_{2}C_{6}H_{4}Me-4)O]Cl(NO)(PPh_{3})_{2}$. A solution of Os[CH₂CH₂)Cl(NO)(PPh₃)₂ (0.40 g, 0.50 mmol) in tetrahydrofuran (30 ml) was treated with commercial (Fluka) ethyl diazoacetate, N₂CHCO₂C₂H₅ (2 drops), followed by a solution of tosyliminooxosulphurane in toluene (0.10 mol/ 1, 5.3 ml, 0.53 mmol) and the mixture stirred for 1 h. The solvent was then removed under reduced pressure and the bright yellow residue recrystallised from tetrahydrofuran/ ethanol. Yield 0.44 g (82%). M.p. 194°C.

Acknowledgements

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